

# MultiTek<sup>®</sup> Halides & Sulfur in Liquefied Petroleum Gas (LPG) by Oxidative Pyrohydrolytic Combustion followed by Ion Chromatography Conductivity Detection

- **Fully Automated Combustion System**
- **Halogen Determination**
- Sample Versatility

#### **Keywords:**

MultiTek, pyrohydrolysis, halides, liquefied petroleum gas (LPG), n-butane, fluoride, chloride, bromide, sulfur, combustion ion chromatography (CIC)

# INTRODUCTION

Hydrofluoric acid alkylation is a common technique for the production of high octane gasoline and LPG fractions such as n-butane and propane. Even though acid settlers and adsorbent beds are used to remove fluoride, traces still manage to make it into the final product that is delivered to the consumer.

Butane is a common feed used in production and can be blended with other compounds to achieve a desired product. When the fluorides, chlorides and sulfur in the butane react, they can produce extremely corrosive acids such as hydrofluoric and hydrochloric acids. As a result, gas leaks can evolve from these corrosive effects and pushes the need to monitor concentrations of halides and sulfur.

This procedure analyzes total halogens and sulfur compounds (F/Cl/Br/SO<sub>4</sub>) in liquefied petroleum gases. Specifically n-butane was chosen, but heavier samples such as pentane and hexane are applicable. The process starts with pyrohydrolysis of the samples at 1050°C. In the pyrotube with the help of steam, halogen containing compounds are converted into an acid gas state. Sulfur containing compounds are oxidized to sulfur dioxide (SO<sub>2</sub>) and in a unequalled occurrence eventually converted to sulfite (SO<sub>3</sub>) and sulfate (SO<sub>4</sub>). Once combustion is completed, gases are condensed and absorbed in a solution composed of UHP 18.2 MOhm water and phosphate internal standard. The solution is then transferred by the MultiTek® to an injection system on the Ion Chromatograph (IC) for conductivity analysis.

# REACTIONS

- Halogen Reaction
- $R X_{(F^{-}Cl^{-}Br^{-})} \xrightarrow[1050^{\circ}C + 0_{2} + H_{2}O]{} H X_{(F^{-}Cl^{-}Br^{-})}$
- Sulfur Reaction

 $R - S \xrightarrow[1050^{\circ}C + O_2 + H_2 0]{} SO_2 \xrightarrow[scrubber solution]{} SO_3^{2-} + SO_4^{2-}$ 

To ensure the highest oxidization conversion from sulfite to sulfate, optima grade hydrogen peroxide can be added to the scrubber solution.

### EXPERIMENTAL CONDITIONS

### Instrumentation

Antek MultiTek Horizontal, Antek Model 740 boat inlet system, Antek Model 735 syringe drive, Antek Model 734 Gas/LPG Sampler and suppressed IC system.



### Instrument Parameters

734 Sample Loop (µl)	15
IC Sample Loop (µl)	250
GFC1- Ar/He (ml/min)	130
GFC2- Pyro O2 (ml/min)	450
GFC4- Carrier O2 (ml/min)	30
734 Carrier- Ar or He (ml/min)	20
Furnace (°C)	1050
Sample Burn Time (mm:ss)	02:45



# **APPLICATION NOTE**



## RESULTS

 Repeatability of ~25 ppm (w/w) n-butane (n=6)

	Fluoride	Chlorid e	Bromid e	Sulfur
AVG	22.3	24.4	24.8	23.6
STD	0.23	0.57	0.44	1.04
RSD	1.03%	2.33%	1.78%	4.39%

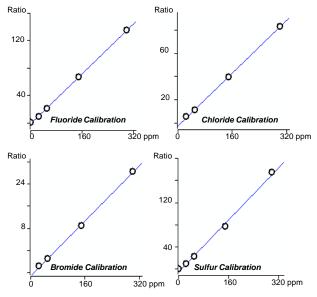
### Limit of Detection

	Fluoride	Chlorid e	Bromid e	Sulfur
Limit of Detectio n	374 ppb	164 ppb	46 ppb	1 ppm

### Standards

IC calibration was performed through entire sample flow path using certified n-butane matrix standard materials (0-300ppm). The compounds used for halide addition are fluorobenzene, chlorobenzene, bromobenzene and dimethyl sulfide in n-butane.

### Calibration



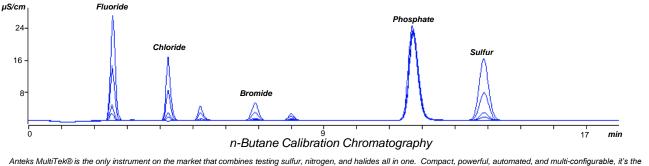
# Correlation Detection's linearity extends over a dynamic range of 10<sup>3</sup>.

0 – 300 ppm	Fluoride	Chlorid e	Bromid e	Sulfur
Correlation Coefficient	0.999	0.999	0.999	0.998
Curve Fit	Linear	Linear	Linear	Linear

CONCLUSION

The results demonstrate that the MultiTek Analyzer coupled with Ion Chromatography provides a sensitive, automated and reliable analysis of halides and sulfur in liquefied petroleum gas. This analysis will allow the refinery and other processes to monitor concentration levels to improve plant safety and efficiency.

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